



DEPARTMENT OF MECHANICAL ENGINEERING
LESSON PLAN

Course Code	Course Title	Semester	Branches	Conduct Periods /Week	A.Y	Date of commencement of Semester
20ME3T04	ENGINEERING THERMODYNAMICS	III	Mechanical Engineering	6	2021-22	25 -10-2021

COURSE OUTCOMES

- 1 Apply basic principles and Zeroth law of thermodynamics to solve problems. [K3]
- 2 Apply first law of thermodynamics to different thermodynamic systems. [K3]
- 3 Apply second law of thermodynamics and general thermodynamic property relations to solve Problems. [K3]
- 4 Describe the thermodynamic concepts of pure substances and identify their properties using Standards. [K2]
- 5 Analyse various power cycles, vapour power cycles and Refrigeration cycles. [K4]

UNIT	Out Comes/ Blooms Level	Topics No.	Topics/Activity	Text Book /Reference	Conduct Hour	Delivery Method	
I	CO1: Apply basic principles and Zeroth law of thermodynamics to solve problems. [K3]	1. Introduction: Basic Concepts					
		1.1	Thermodynamic Systems, Types of Systems, Control volume.	T ₁ & T ₂	1	Chalk, Talk, & videos	
		1.2	Macroscopic and Microscopic Viewpoints, Thermodynamic Equilibrium	T ₁ &T ₂	1		
		1.3	Property and Types, State, Process, Cycle	T ₁	1		
		1.4	Reversible Process, Quasi Static Process	T ₁ & R ₁	1		
		1.5	Irreversible Process, Energy in State and in Transition- Types, Work and Heat	T ₁ & R ₁	1		
		1.6	Point Function and Path Function	T ₂ & R ₂	1		
		Zeroth Law of Thermodynamics					
		1.7	Definition, Concept of Temperature	T ₁ &T ₂	1		
		1.8	Principles of Thermometry – Reference Points	T ₁ & T ₂	1		
		1.9	Constant Volume gas Thermometer	T ₁ , & R ₁	1		
		1.10	Electrical Resistance Thermometer, Thermocouple, Ideal Gas Scale	T ₁	1		
	C.B.S-1	1.11	Constant pressure gas thermometer	T ₁	1		
Total					11		



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		2. Ideal Gas Laws:			
II CO2: Apply first law of thermodynamics to different thermodynamic systems. [K3]	2.1	Definitions- Equation of State, Universal Gas Constant.	T ₁ &T ₂	1	Chalk, Talk, & videos
	2.2	Specific Heat at Constant Volume& Pressure	T ₁ & R ₁	1	
	2.3	Principle of First law of thermodynamics- Joule's Experiment, Internal Energy	T ₁ & R ₁	1	
	2.4	Enthalpy , PMM-1	T ₁ & R ₁	1	
	2.5	First law applied for a Non Flow Processes for constant volume and pressure process	T ₁ & R ₁	1	
	2.6	Isothermal process	T ₁ & R ₁	1	
	2.7	Adiabatic and polytropic process	T ₁ & R ₁	1	
	2.8	First Law Applied to a Flow System - Steady Flow Energy Equation	T ₁ & R ₁	1	
	2.9	SFEE Applications	T ₁ & R ₁	1	
	2.10	Throttling Process and Free Expansion	T ₁ & R ₁	1	
	2.11	problems	T ₁ & R ₁	2	
	C.B.S-2	2.12 Vander wall's equation of state.	T1	1	
Total				13	
		3. Second Law of Thermodynamics			Chalk, Talk &PPT
CO3: Apply second law of thermodynamics and general thermodynamic property relations to solve Problems. [K3]	3.1	Limitations of the 1 st Law of Thermodynamics	T ₂ & R ₂	1	
	3.2	Thermal Reservoir, Heat Engine, Refrigerator and Heat pump.	T ₂ & R ₂	1	
	3.3	Kelvin-Planck and Clausius Statements	T ₁ & T ₂	1	
	3.4	Equivalence Kelvin-Planck and Clausius Statements & PMM-2	T ₁ & T ₂	1	
	3.5	Carnot Cycle	T ₁ & T ₂	1	
	3.6	Carnot's theorem and its corollaries	T ₁ & T ₂	1	
	3.7	Thermodynamic scale of Temperature		1	
	3.8	Clausius Inequality,	T ₁ & T ₂	1	
	3.9	Entropy, Principle of Entropy Increase	T ₁ & T ₂	1	
	3.10	Availability and Irreversibility	T ₁ & T ₂	1	
	3.11	Thermodynamic Potentials - Gibbs and Helmholtz Functions	T ₂ & R ₂	1	
	3.12	Maxwell Relations	T ₂ & R ₂	1	
	3.13	Elementary Treatment of the 3 rd Law of TD	T ₂ & R ₂	1	
	3.14	problems	T1	2	
			Total	15	



		4. Properties of Pure Substances					
IV	CO4: Describe the thermodynamic concepts of pure substances and identify their properties using Standard s. [K2]	4.1	Definitions- Sensible heat, Latent heat and total heat	T ₁ & T ₂	1	Chalk, Talk, & Tutorials	
		4.2	Phase Transformation- Formation of Steam- P-V diagram, P-T diagram	T ₁ & T ₂	1		
		4.3	T-S diagram and h-s diagram or Mollier Chart.	T ₁ & T ₂	1		
		4.4	PVT Surface, Triple point and critical point	T ₂ & R ₁	1		
		4.5	Dryness Fraction, Steam Calorimetry	T ₁ & R ₁	1		
		4.6	Property Tables	T ₁ & R ₁	1		
		4.7	Various Thermodynamic Processes like constant volume and pressure process	T ₁ & T ₂	1		
		4.8	Isothermal process	T ₁ & T ₂	1		
		4.9	Adiabatic and polytropic process	T ₁ & T ₂	1		
		4.10	problems	T ₁ & T ₂	2		
	C.B.S-3	4.11	Compressed Liquid	T ₁ & T ₂	1		
		Total			12		
V	CO5: Analyse various power cycles, vapour power cycles and Refrigeration cycles. [K4]	5. Power Cycles:					
		5.1	Otto Cycle - P-V diagrams & T-S diagrams Thermal Efficiency, Mean Effective Pressure	T ₁ & T ₂	1	Chalk, Talk, & Tutorials	
		5.2	Diesel Cycle	T ₁ & T ₂	1		
		5.3	Dual Combustion cycle	T ₁ & T ₂	1		
		5.4	Comparison of Cycles	T ₁ & T ₂	1		
		5.5	Problems	T ₁ & T ₂	3		
		Vapour Power & Refrigeration Cycles					
		5.6	Brayton Cycle - P-V diagrams & T-S diagrams- Thermal Efficiency	T ₂ & R ₁	1		
		5.17	Ideal Rankine Cycle - P-V diagrams & T-S diagrams - Thermal Efficiency	T ₂ & R ₁	1		
		5.8	Bell- Coleman Cycle - COP	T ₂ & R ₁	1		
		5.9	VCR Cycle -COP	T ₂ & R ₁	1		
		5.10	Simple problems	T ₂ & R ₁	2		
		Total			13		
Cumulative Proposed Periods					64		

Where : **C.B.S** = Content Beyond the Syllabus



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Text Books:

S.N o.	Authors, Book Title, Edition, Publisher, Year of Publication
T1	P.K Nag, Engineering Thermodynamics, 6 th Edition, McGraw-Hill Publication, 2020
T2	Y.A.Cengel & M.A.Boles, Thermodynamics, 9 th Edition – McGraw-Hill Publication, 2019

Reference Books:

S.N o.	Authors, Book Title, Edition, Publisher, Year of Publication
R1	R.K Rajput, A Textbook Of Engineering Thermodynamics, 5th Edition, Lakshmi Publication, 2016.
R2	P.Chattopadhyay, Engineering Thermodynamics, 2 th Edition, Oxford Higher Edn Publ, 2015

Web Details

W1	https://youtube.com/playlist?list=PLD8E646BAB3366BC8
W2	https://sites.google.com/site/thermojan2019/
W3	https://www.researchgate.net/deref/http%3A%2F%2Fwww.thermalfluidscentral.org%2F

S.NO.	Details	Name	Signature
i.	Faculty	Mr. B SRINIVAS	
ii.	Faculty II (for common Course)	Mr. G VEERENDRA KUMAR	
iii.	Course Coordinator	Mr. B SRINIVAS	
iv.	Module Coordinator	Dr. R. LALITHA NARAYANA	
v.	Program Coordinator	Dr. A. GOPI CHAND	

Principal